

LETTER TO THE EDITOR

Diverse Functional Components from *Robinia Pseudoacacia* Flowers by High-efficient Extraction

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Robinia pseudoacacia is one of the main tree species for greening in Henan Province, while *R. pseudoacacia* flowers (RPFs) are not enough utilized except edibility. The active ingredients in solution extracts of *Robinia Pseudoacacia* flowers (RPFs) were detected and analyzed. The results for the RPFs characteristic groups indicated that there are organic compounds and contain double bonds in RPFs, which may contain saturated hydrocarbons and unsaturated hydrocarbons. The results on non-volatile substances and volatile substances in RPFs indicated that the extracts of acetone, benzene and ethanol are rich in active substances. The findings suggested that RPFs have great potential in high-efficient extraction resourcing for biomedicine and chemical materials.

I Introduction

Robinia pseudoacacia were introduced to Asia, Europe and other regions, and it has been widely cultivated in the world. The height of the *R. pseudoacacia* can reach 25 meters, which can form a good garden landscape. At the same time, due to its low environmental requirements and certain anti-pollution ability, it has been planted in many industrial areas (Liu et al. 2015, Shen et al. 2017).

The *R. pseudoacacia* flowers (RPFs) contain oleic acid, enamel and flavonoids and it is an excellent honey plant. The production of RPFs increased dramatically and cannot be utilized, which has caused great waste (Eriksson et al. 2018, Kumar et al. 2019). In this study, in order to understand the effective active ingredients and relative contents of RPFs, the characteristic groups of RPFs were detected by Fourier Transform infrared spectroscopy (FTIR). In order to know the composition of volatile substances in RPFs, the RPFs raw powder was extracted with three solutions of benzene, acetone and ethanol, and the extracts of the three solutions were detected by Gas Chromatography-Mass Spectrometer (GC-MS/MS) (Cheng et al. 2018). At the same time, quadrupole time of flight-liquid chromatography-mass spectrometry (QTOF-LC-MS) helps analyze highly polar, hardly volatile, thermally unstable compounds that cannot be analyzed by gas chromatography-mass spectrometry (GC-MS/MS) (Chen et al. 2015, Zheng et al. 2015).

II Materials and Methods

Materials

The RPFs were picked large quantity from *Robinia pseudoacacia* (Wu et al. 2014, Zhao et al. 2016). In mid-April The RPFs were cleaned and placed at room temperature (25°C) for 24 h.

Methods

Extraction. RPFs powder were separately extracted by ethanol, benzene, or acetone with an automatic FOSS Soxhlet Extracted apparatus (Agilent, USA) for 3-5 h at 80°C, 60°C, and 56°C, respectively, then extraction liquids were evaporated between 45°C-50°C under 0.01 MPa vacuum, and ultimately concentrated to 10 mL.

FT-IR analysis. Previously prepared and dried RPFs powder was baked under an infrared lamp to reduce the absorption of moisture in the air and ensure the dryness of the sample (Lakra et al. 2015; Lam et al, 2019).

GC-MS/MS determination. the GC conditions in a quartz capillary column (30 mm × 0.25 mm × 0.25 μm) were start at 50°C, no retention, reach 130°C at a continual heating rate of 5 °C/min, heat to 180°C at 2 °C/min, and then immediately switch to a heating rate of 30 °C/min to.

QTOF-LC-MS. column ACQUITY UPLC® BEH C18 1.7 μm × 2.1 × 50 mm, capillary (kV) achieve a temperature of 300°C for 5 min (Henninot et al. 2015, Kayacelebi et al. 2014)

III Results

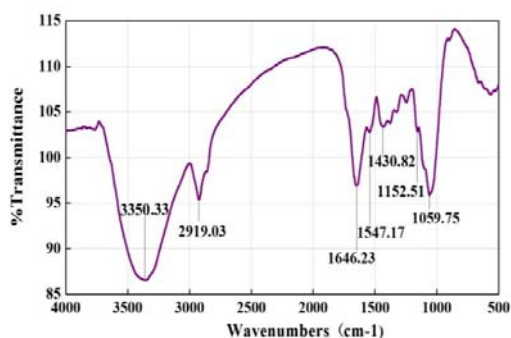
(1) Group change characteristics of RPFs extracts

Fourier transform infrared (FTIR) spectroscopy is a high-sensitivity rapid detection technology that can quickly and efficiently identify chemical bonds and functional groups of many kinds of compounds, and many scientific studies use it to make preliminary determinations of substances (Gupta et al. 2015). The extract was obtained by distillation, filtration and concentration of three solvents of benzene, acetone and ethanol, and then the extract and the original solid powder were detected by FT-IR, in order to reveal the change of chemical groups in the extract (Meng et al. 2015).

Seven distinct peaks were detected in the original powder of RPFs (Figure 1). Significant absorption peaks appeared in the range of 4000-2500 cm⁻¹, indicating the presence of X-H bond. Due to the small amount of water in the sample, it is concluded that the absorption peak appearing around 3350 cm⁻¹ may be caused by the stretching vibration of the O-H bond (Steinborn et al. 2016). The absorption peak appearing at 2919 cm⁻¹ may be caused by the stretching vibration of the C-H bond.

In summary, it can be concluded that organic compounds are present in the RPFs and saturated hydrocarbons may be present. In addition, by analyzing the infrared spectrum of the three solution extracts, it was found that ethanol as the extract has the best effect on the original powder extraction of RPFs.

a)



b)

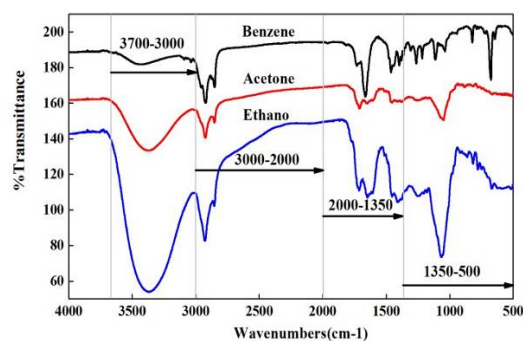


Figure 1: Figure a) is an infrared spectrum of the RPFs raw powder, and Figure b) is an infrared spectrum of the three solution extracts. From the Figure a, we can see the organic compounds contained in the RPFs. The infrared spectrum of the three solution extracts (Figure b) shows that ethanol as the extract has the best effect on the original powder of RPFs.

(2) Volatile components in extracts of RPFs

There are many volatile components shows a variety of bioenergy and biologically active. The extract was originally used for liquid bioenergy development and the remainder of bioactive component resources. Figure 2 a shows the percentage of benzene solution extracts ingredients in chemical raw materials, bioenergy, biomedicine, spices, food additives, and cosmetics. 1-Hexanol, 2-ethyl- is an important chemical raw material, in addition to being used in the production of plasticizers. (Singh and Bukowska 2015, Yuan et al. 2016).

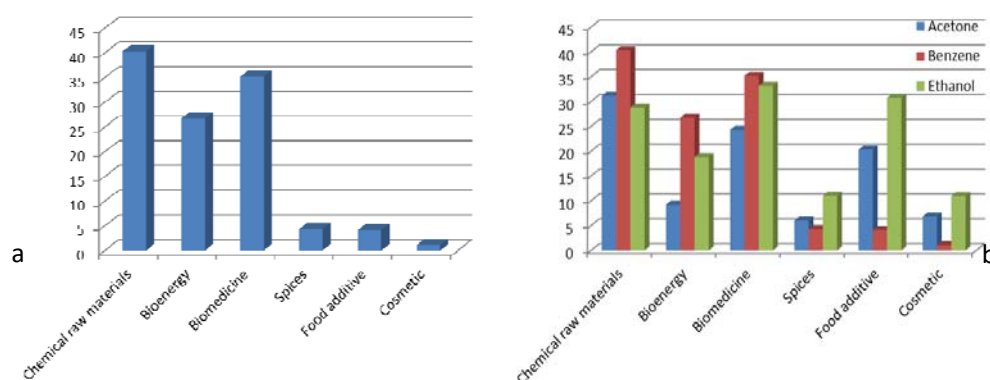


Figure 2: a) The functional classification of benzene solution extract. b) Comparison of the functional classification of substances in different solution extracts.

(3) Non-volatile components in extracts of RPFs

There are many non-volatile components detected in the three different solvent extracts of *Robinia pseudoacacia* via QTOF-LC-MS. As can be seen from the Figure 2 b, the chemical raw materials, bioenergy and biomedicine extracted from the benzene solution are the most. And from ethanol extracts, the most spices, food additives and cosmetic raw materials can be obtained (Hao et al. 2015, Cai et al. 2016). By comparison, it can be found that in the three extraction solutions, the extraction effect of acetone is the worst overall. In the production process, we can consider abandoning acetone as the extraction solution.

Many active substances were detected in the benzene solution extract. And these substances mainly can be used for the production of biomedicine, spices, chemical raw materials, food additives and cosmetics. The main ingredients of biomedicine are: Arachidonic acid, Vitamin K2 and Ethyl linoleate (He et al. 2015).

There are many chemicals includes ginkgolic acid, rhynchophylline and vitexin. Ginkgolic acid has an antibacterial effect and it inhibits the growth of *Mycobacterium tuberculosis* in vitro (Lu et al. 2014, Tsikas et al. 2014). Rhynchophylline has antihypertensive effect on renal hypertension, and clinical trials have certain curative effect on patients with stage I and II hypertension. Emodin has a significant diuretic effect and it has anti-tumor effect on renal medulla $\text{Na}^+\text{-K}^+\text{-ATPase}$. At a certain concentration, emodin can reduce the maximum growth density of human lung cancer A-549 cells and decrease the division index.

IV Conclusion

There exist many active ingredients in the extract of RPFs that can be used as biomedicine, chemical raw materials, cosmetics, spices, food additives, etc. (Ryu et al. 2015). Although the types of substances are diverse, the content of a substance is very small. What's more, the cost of separating these active ingredients separately is high, which has become a major obstacle to how to apply the active ingredients in the hedgehog RPFs to actual production. However, we can think about the problem from another perspective. If we can separate the active ingredients of a certain species and the substances that inhibit this effect, it may be able to meet the actual demand while reducing the certain production costs. Therefore, in the following period, we will make further experiments and explorations on how to effectively use the ingredients in the RPFs, and we also hope that all of us can find a solution to the problem as soon as possible.

Acknowledgements

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